

Last Time

① Light is made of discrete packets - Photons

$$\boxed{E = hf}$$

$$hc = 1240 \text{ eV/nm}$$

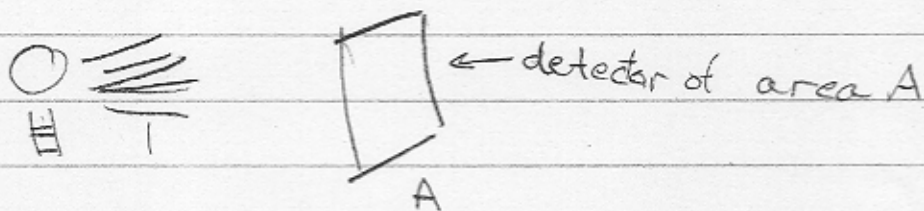
$$2 \text{ eV} \sim 600 \text{ nm} \sim \text{red}$$

Visible light \approx a couple of eV's

② When lots of photons are measured see the wavelike nature of light

a) -- see slides

b) A relation between the wavelike and Particle like properties:



Wave

energy absorbed

area

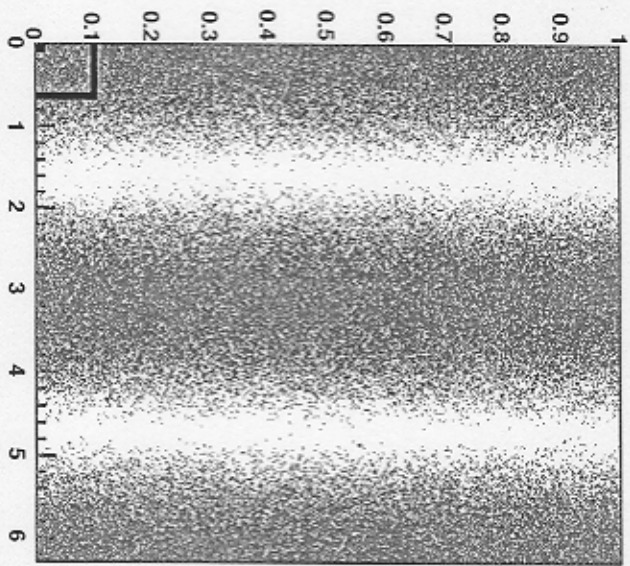
time

$$\Delta E = I A \Delta t$$

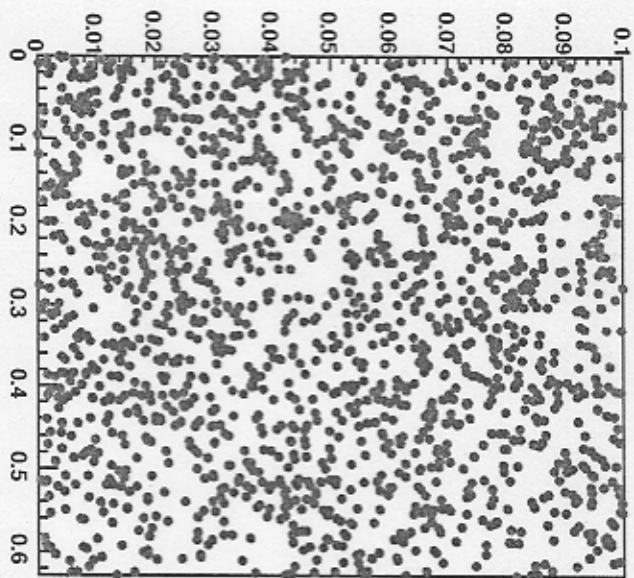
Intensity of light $I = \langle \vec{S} \rangle \propto \vec{E} \times \vec{B} \propto E^2$

Counting Photons

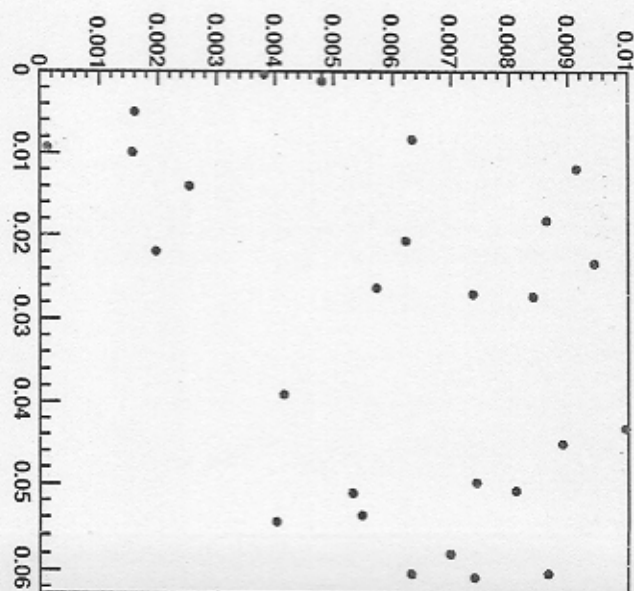
Big Detector



Medium Detector



Small Detector



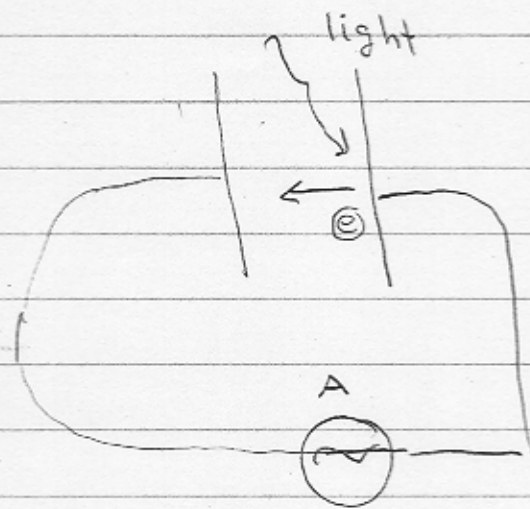
Particle Picture

$$\begin{array}{ccccc} \Delta E & = & \Delta N & hf & \\ \uparrow & & \uparrow & \nwarrow & \\ \text{energy} & & \text{number} & & \text{Energy per} \\ \text{absorbed} & & \text{absorbed} & & \text{photon} \end{array}$$

So,

$$\Delta N \propto \frac{I A \Delta t}{hf}$$

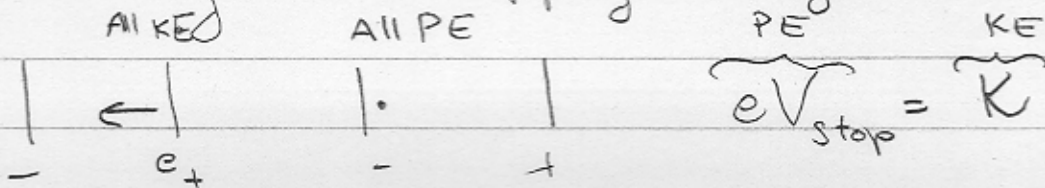
③ Photo-Electric Effect - Atoms are small enough to see the discrete nature



$$K = hf - \phi$$

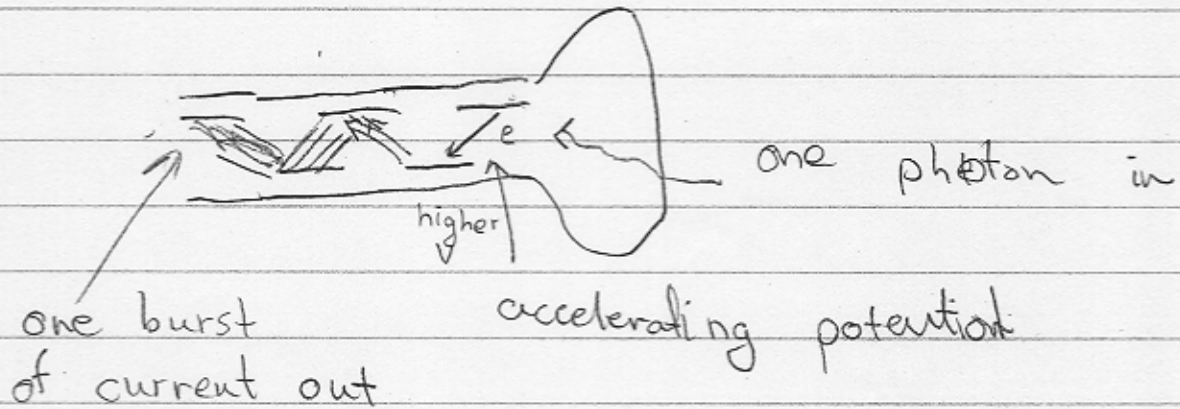
\uparrow KE of ejected electron $\sim eV$
 \uparrow energy of light $\sim eV$
 \nwarrow energy required to strip of electron $\sim eV$

One can measure the electron's energy by turning a stopping voltage



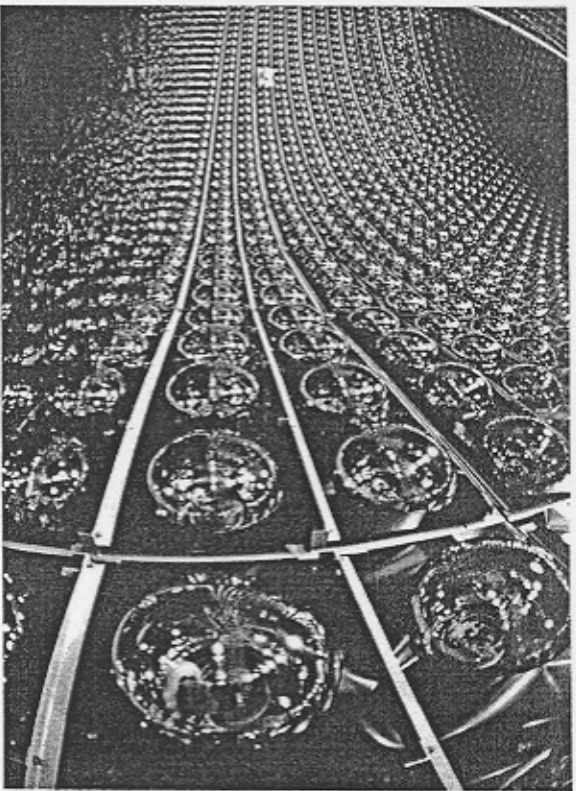
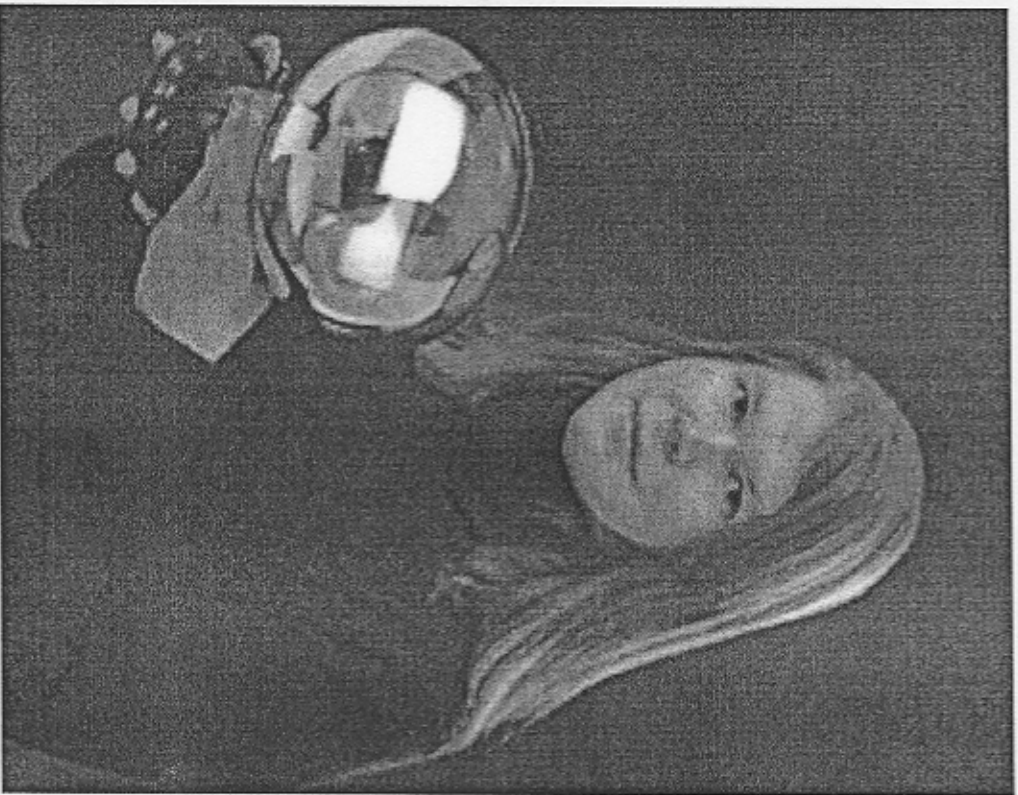
Practical Uses of the Photo-electric effect :

Photomultiplier Tube :

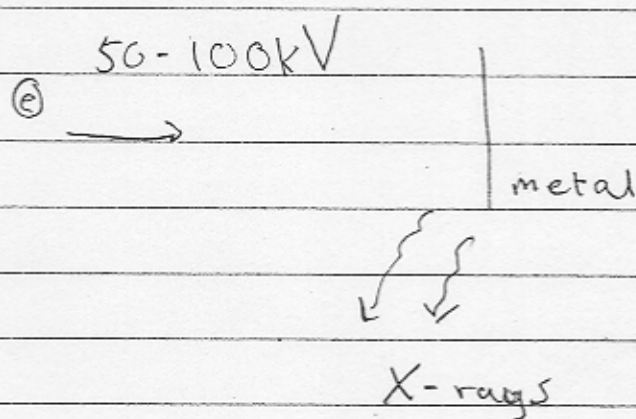


Photomultiplier Tubes

(Janet Conrad – MIT professor, co-spokesperson for the MiniBooNE Experiment)



X-rays



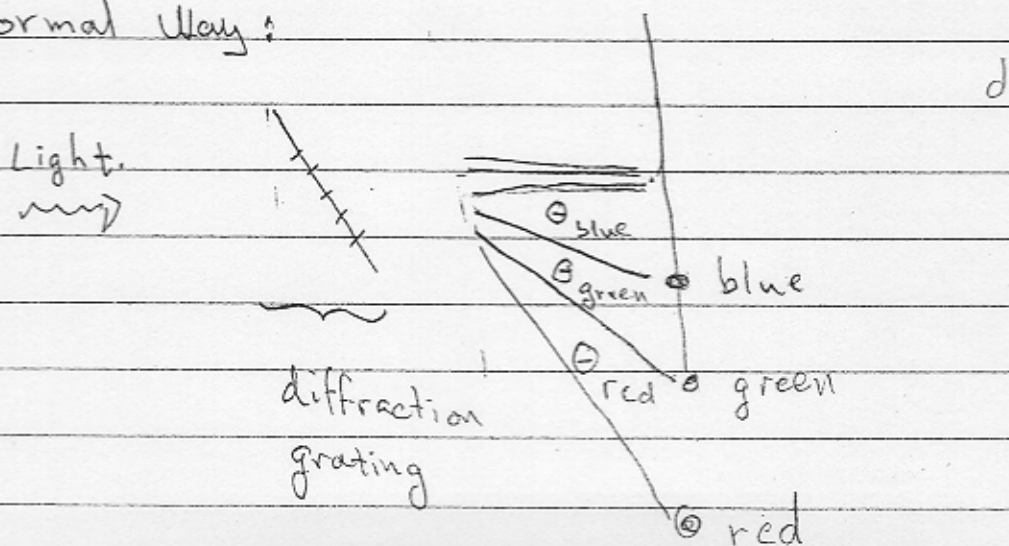
• Very high frequency $\lambda_x \sim 10^{-10}\text{ m}$

- Compare $\lambda_{\text{visible}} \sim 6000 \times 10^{-10}\text{ m}$

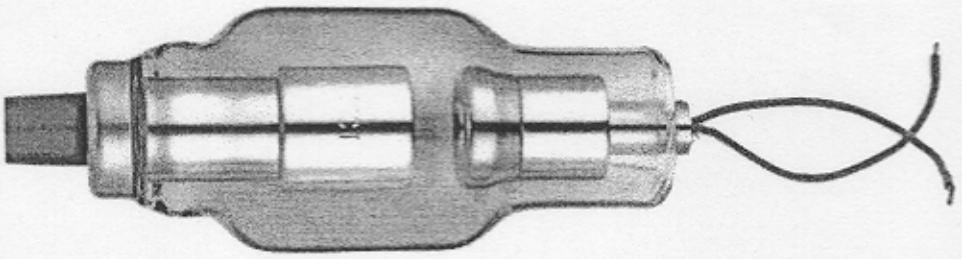
• $E = hf$ is a big step! This light is not continuous on human scales

• Problem how to measure the wavelength?

Normal Way:

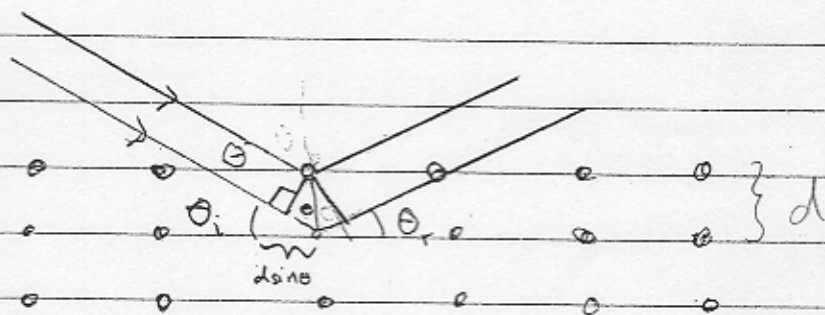


X-ray Tube – Used for security. 160 Killo Volts



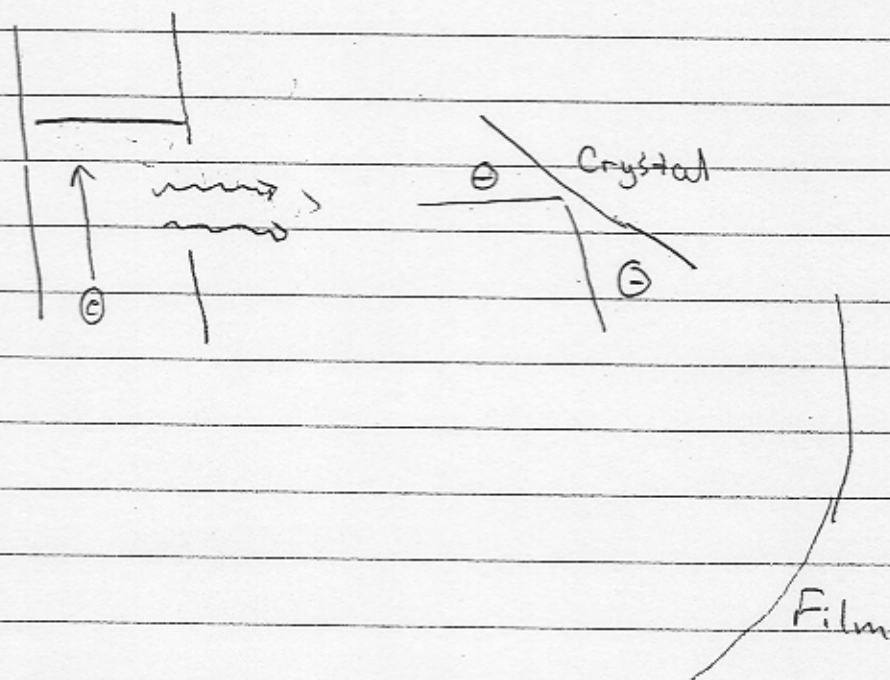
How to make a diffraction grating where the spacing is $\sim \lambda$.

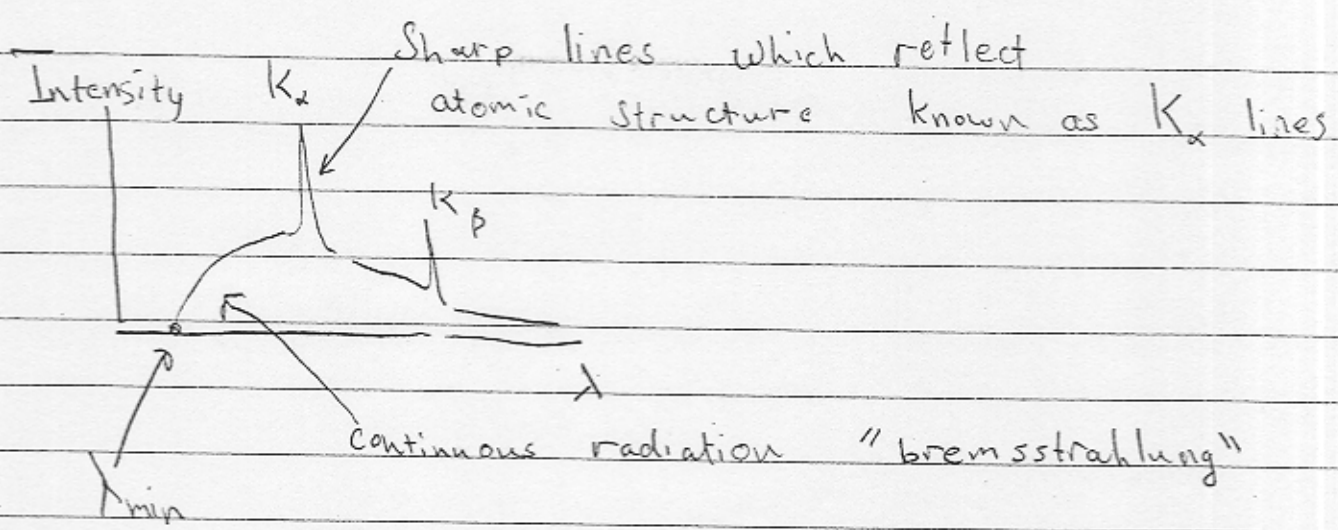
Answer: Make a diffraction grating out of a crystal (Bragg, von Laue)



$$2d \sin \theta = n\lambda \quad n = 1, 2, 3, 4, \dots$$

Result



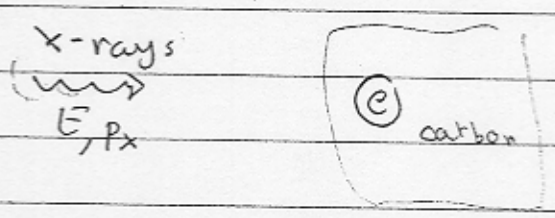


λ_{min} is determined when all of the energy of the electron is converted to a single photon

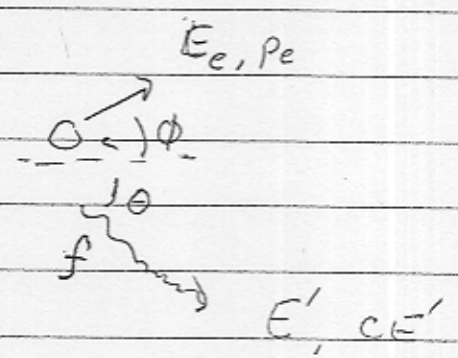
$$eV = hf = \frac{hc}{\lambda_{min}}$$

The Compton Process (1922)

Before:



After:

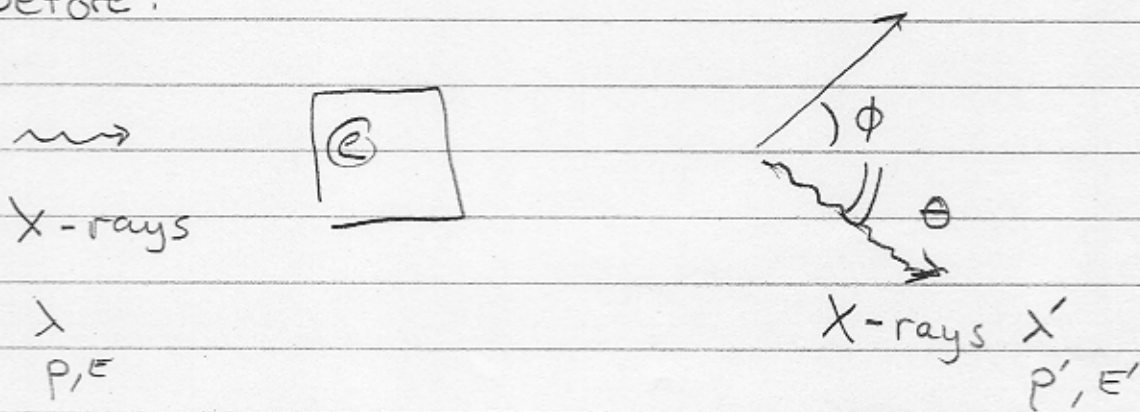


Use relativistic energy and momentum conservation and

$$E = hf$$

The Compton Process

Before:

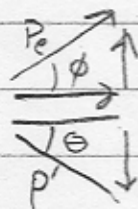


Particle Description

p-conservation:

$$(2) \quad p = p' \cos \theta + p_e \cos \phi$$

$$(3) \quad 0 = -p' \sin \theta + p_e \sin \phi$$



E-conservation

$$(1) \quad \underbrace{E}_{cp} + m_e c^2 = \underbrace{E'}_{cp'} + \underbrace{E_e}_{\sqrt{(cp_e)^2 + (m_e c^2)^2}}$$

Knowns:

θ, p

Unknowns:

$\phi, p_e, p' \Rightarrow$ Use equations (1), (2), (3) solve for ϕ, p_e, p' in terms of p, θ

Algebra: (See handout)

$$(*) \quad \frac{1}{m_e c} (1 - \cos\theta) = \frac{1}{p'} - \frac{1}{p}$$

Now assume

$$E = cp = \frac{hc}{\lambda} \quad \text{and} \quad E' = \frac{hc}{\lambda'}$$

$$p = \frac{h}{\lambda} \quad \text{and} \quad p' = \frac{h}{\lambda'}$$

Then multiplying (*) by h

$$\frac{h}{m_e c} (1 - \cos\theta) = \frac{h}{p'} - \frac{h}{p}$$

Or

$$\frac{h}{m_e c} (1 - \cos\theta) = \lambda' - \lambda$$

Energy (Algebra Skipped if pressed for time)

$$(1) \quad E + m_e c^2 = E' + E_e$$

Momentum

$$(2) \quad p = p' \cos \theta + p_e \cos \phi$$

$$(3) \quad 0 = -p' \sin \theta + p_e \sin \phi$$

Also have, $E = cp$, $E' = cp'$, $E_e = \sqrt{(cp_e)^2 + m_e c^2}$

Algebra: use the 1 - equation to eliminate p_e , ϕ

$$p' \sin \theta = p_e \sin \phi \quad (2)$$

$$(p - p' \cos \theta) = p_e \cos \phi \quad (3)$$

Squaring both equations and add:

$$p^2 - 2pp' \cos \theta + p'^2 \cos^2 \theta = p_e^2 \cos^2 \phi$$

$$p'^2 \sin^2 \theta = p_e^2 \sin^2 \phi$$

$$p^2 - 2pp' \cos \theta + p'^2 = p_e^2 \quad \Leftarrow \text{so far used one equ to elim } \phi$$

Ripped

Now use E -consue

$$(cp - cp') + m_e c^2 = E_e$$

$$[(cp - cp') + m_e c^2]^2 = E_e^2$$

$$(cp - cp')^2 + 2(cp - cp')(m_e c^2) + (m_e c^2)^2 = E_e^2$$

$$(cp - cp')^2 + 2(cp - cp') m_e c^2 = E_e^2 - m_e c^2$$

$$(cp - cp')^2 + 2cp - 2cp' + m_e c^2 = (cp_e)^2$$

$$(p - p')^2 + 2(p - p') m_e c = p_e^2$$

Putting in:

$$p^2 - 2pp' \cos \theta + p'^2 = (p - p')^2 + 2(p - p') m_e c$$

Working:

$$\frac{1}{m_e c} (1 - \cos \theta) = \frac{1}{p'} - \frac{1}{p}$$

Now assume: $p = h\nu$ $p' = h\nu'$

$$\frac{1}{m_e c} (1 - \cos \theta) = \frac{1}{h\nu'} - \frac{1}{h\nu}$$

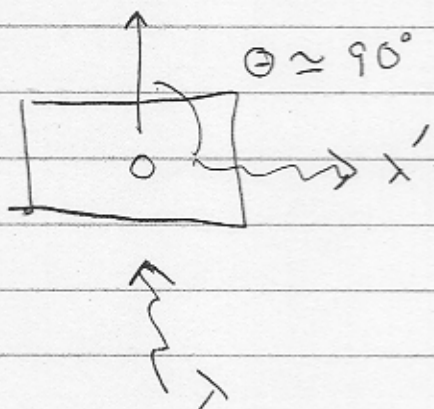
Important Points

① Used Relativity and Used

$$E = cp = hf$$

and it worked

② The Compton wavelength



λ' is longer than λ :

$$\frac{h}{m_e c} (1 - \cos\theta) = \lambda' - \lambda$$

$$\frac{h}{m_e c} = \lambda' - \lambda$$

The wavelength is longer by the Compton wavelength

$\lambda_c \equiv \frac{h}{m_e c} =$ the intrinsic size of the electron

$$\lambda_c = \frac{h}{m_e c} = \frac{hc}{m_e c^2} = \frac{1240 \text{ eV nm}}{0.511 \times 10^6 \text{ eV}} \approx 0.025 \text{ \AA} = 0.0025 \text{ nm}$$

$\lambda \approx \frac{1}{40}$ the size of an atom

③ Why use X-rays
for $\theta = 90^\circ$

$$\lambda_c (1 - \cos\theta) = \lambda' - \lambda$$

$$\lambda_c = \lambda' - \lambda \Rightarrow \lambda' = \lambda + \lambda_c$$

For $\lambda \sim 500 \text{ nm}$ (visible light)

$$\lambda' \approx 500 \text{ nm} + 0.0024 \text{ nm} \approx 500 \text{ nm}$$

↙ small shift

Thus it was essential to use X-rays

④ What is the energy ^{+ speed} of the recoiling electron when the incoming X-ray has energy $\lambda \approx \lambda_c$ at $\theta = 90^\circ$

E-consv $E + m_e c^2 = E' + E_e$

Then

$$\lambda_c (1 - \cos\theta) = \lambda' - \lambda$$

↑
0 for $\theta = 90^\circ$

$\Rightarrow \lambda_c$ for this problem!

$$2\lambda_c = \lambda'$$

$$E = hf = \frac{hc}{\lambda} = \frac{hc}{\frac{hc}{m_e c^2}} = m_e c^2$$

$$E' = hf' = \frac{hc}{\lambda'} = \frac{m_e c^2}{2}$$

$$\text{So } E + m_e c^2 = E' + E_e$$

$$m_e c^2 + m_e c^2 = \frac{m_e c^2}{2} + E_e$$

$$\frac{3}{2} m_e c^2 = E_e \Rightarrow$$

$$\frac{3}{2} m_e c^2 = \gamma m_e c^2 \Rightarrow \gamma = 3/2$$

$$\beta \approx 0.74$$

Thus electrons are pretty relativistic

Q: How important is the binding Energy?