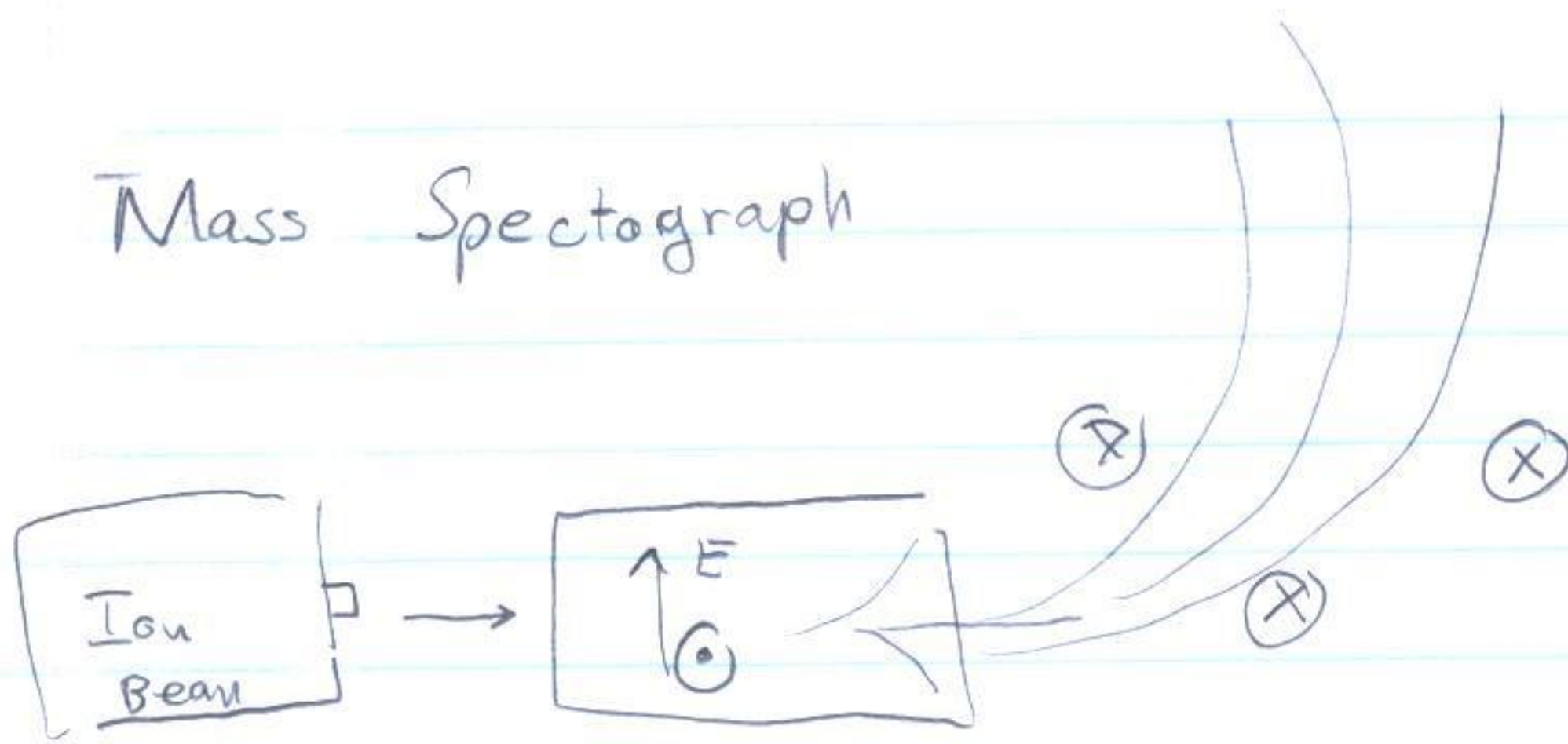


Mass Spectrograph



$$F = qE \quad \text{up}$$

$$F = qvB \quad \text{down}$$

Straight Line :

$$qE = qvB$$

$$E = vB$$

Now we enter a magnetic Field :

$$\frac{mv^2}{R} = qBv$$

$$R = \frac{mv}{qB}$$

$$m = \frac{qRB^2}{v}$$

Doublet method

Compare Two species @ same mass



$$\Delta = [m(C_9H_{20}) - m(C_{10}H_8)]$$

$$\Delta = -12m(H) - m(^{12}C) \quad \leftarrow \text{measured precisely}$$

$$m(H) = \frac{1}{12} [m(^{12}C) + \Delta]$$

$$m(H) = 1.00782503 \pm .00000001$$

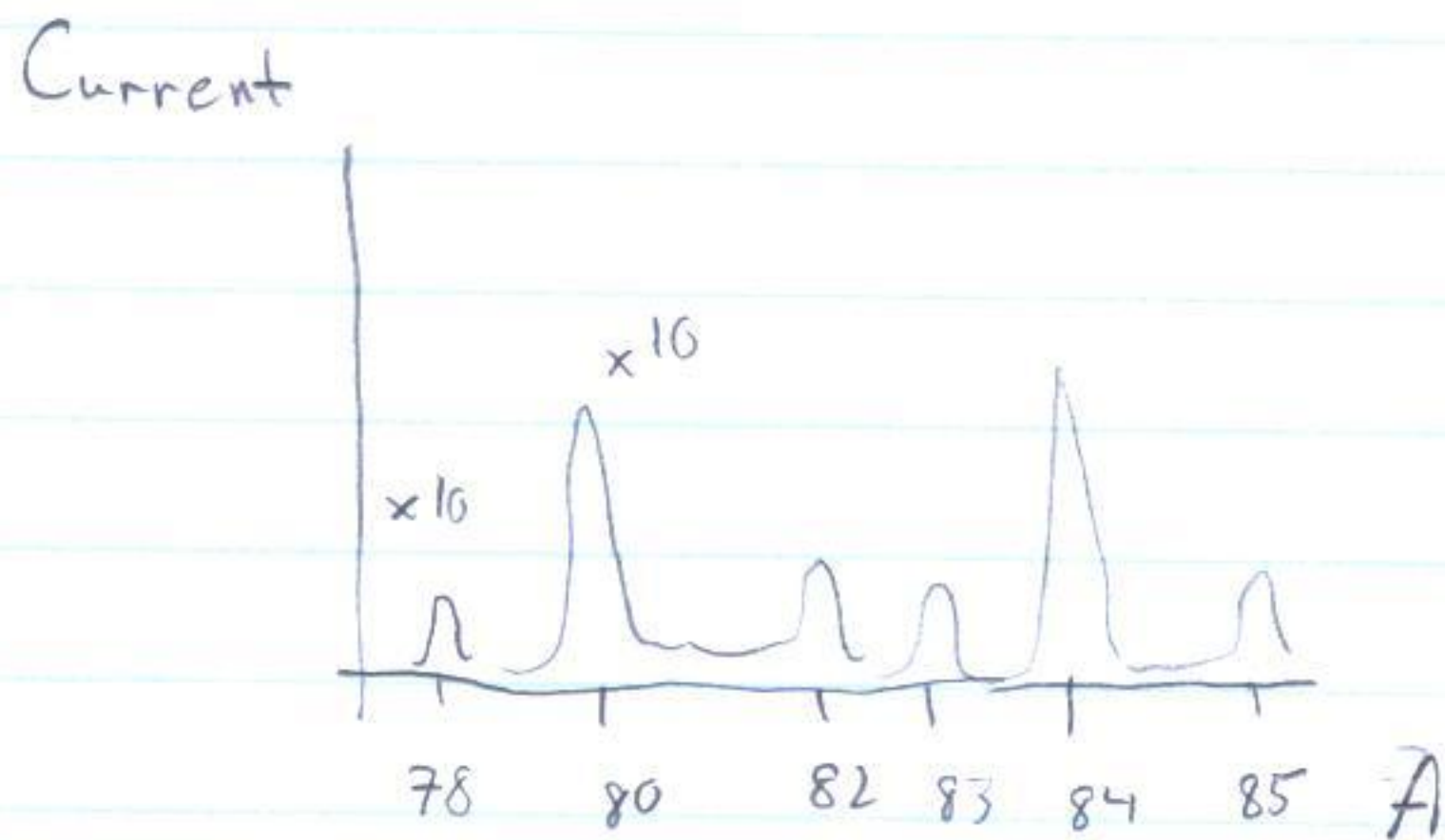
Then to determine say ^{14}N look at C_2H_4 and N_2

$$\Delta' = m(C_2H_4) - 2m(^{14}N) \quad \leftarrow \text{measured precisely}$$

$$m(^{14}N) = m(^{12}C) + 2m(H) - \frac{1}{2}\Delta$$

$$= 14.00307396 \pm 0.00000002$$

The spectro-graph can also be used to measure the relative abundance



$${}^{78}\text{Kr} = 0.356\%$$

$${}^{80}\text{Kr} = 2.27\%$$

$${}^{82}\text{Kr} = 11.6\%$$

$${}^{83}\text{Kr} = 11.5\%$$

$${}^{84}\text{Kr} = 57\%$$

$${}^{86}\text{Kr} = 17.3\%$$

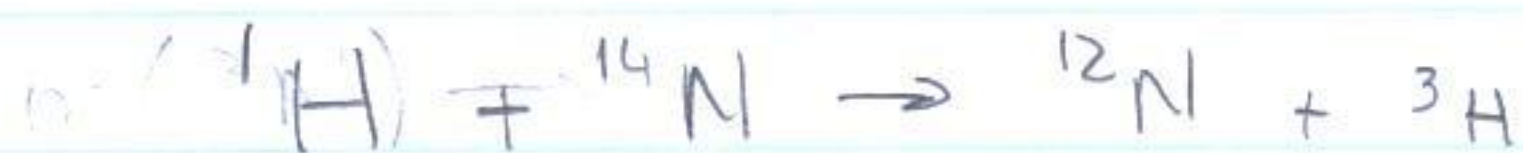
Determining masses of Unstable Nuclei



$$Q = [m(x) + M(X) - m(y) - M(Y)] c^2$$

Measure the kinetic energy of outgoing constituents, and incoming energies

Example:



Measure the kinetic energies $Q = -22 \text{ MeV}$

$$m({}^{12}_7\text{N}) = m({}^1_1\text{H}) + m({}^{14}_7\text{N}) - m({}^3_1\text{H}) - Q$$

Now we can fit the masses

$$B(A, Z) = a_v A - a_s A^{2/3} - a_c \frac{Z(Z-1)}{A^{1/3}} - a_{\text{sym}} \frac{A(1-2Z)^2}{A}$$

↑
↑
↑

Volume $\frac{B}{A}$ Const
0.72 MeV
34 MeV

Then there is a coulomb repulsion:

$$V = k_e Q \left(\frac{1}{R} - \frac{1}{2} \frac{r^2}{R^3} \right)$$

$$U_{\text{electro stat}} = \int 4\pi r^2 dr \rho \left(\frac{k_e q(r)}{r} - \frac{1}{2} \frac{r^2}{R^3} \right)$$

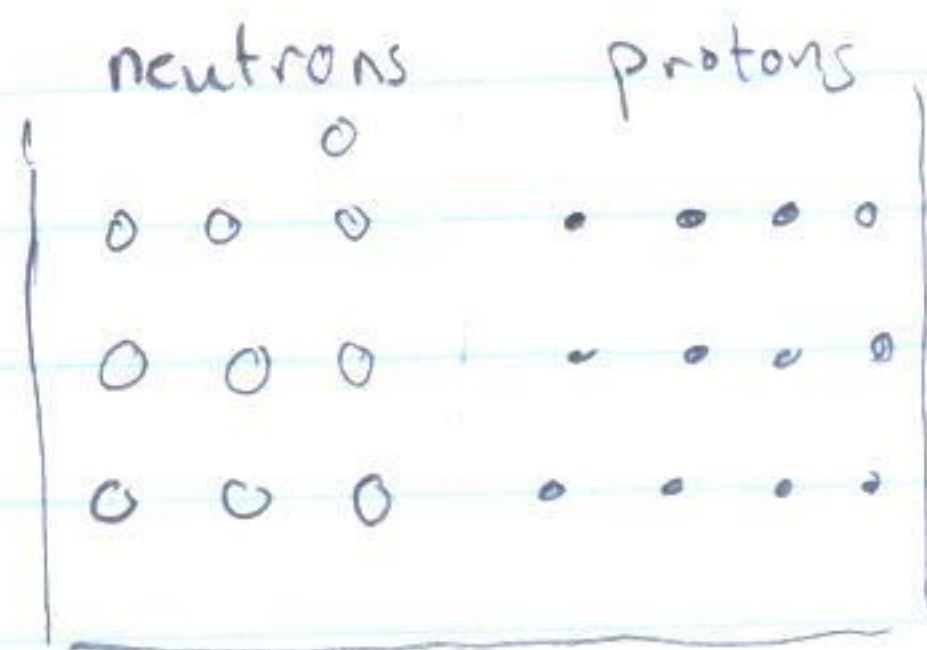
$$= 4\pi k_e \rho \int_0^R \frac{4\pi r^4}{3} \rho dr$$

$$= 4\pi k_e \frac{Q^2}{\left(\frac{4\pi R^3}{3}\right)^2} \frac{4\pi R^5}{3 \cdot 5} = \frac{3}{5} k_e \frac{Q^2}{R}$$

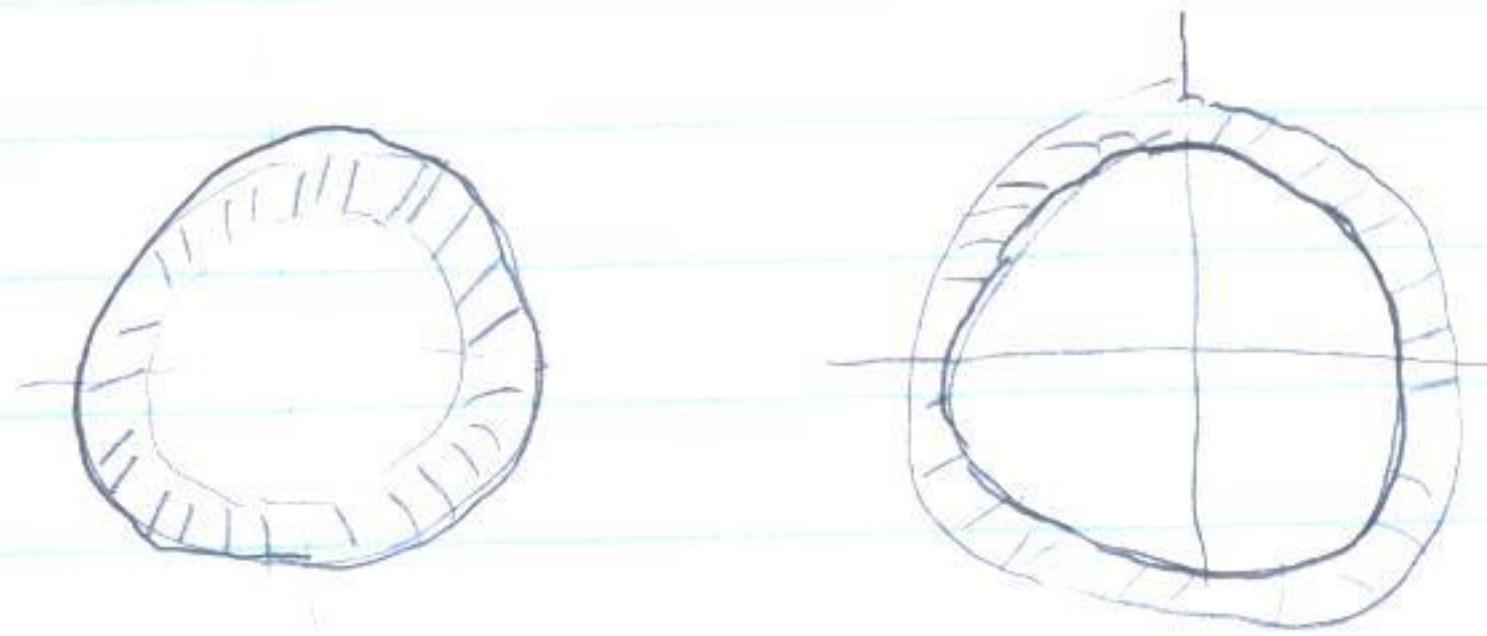
$$U_{\text{elec stat}} = \frac{3}{5} \frac{k_e e^2}{\hbar c} \cdot \frac{\hbar c}{R_A}$$

$$= \frac{3}{5} \underbrace{\alpha_1}_{\frac{1}{137}} \cdot \frac{197 \text{ MeV fm}}{1.2 \text{ fm } A^{1/3}} = 0.72 \text{ MeV}$$

Then consider



If we convert a proton to a neutron then we will add to the fermi sea



$$N = \left(\frac{g}{2\pi^2} \right) \frac{1}{3} \left(\frac{P_f}{\hbar} \right)^3 \cdot V \quad E = \frac{g}{2\pi^2} \frac{1}{3} \left(\frac{P_f}{\hbar} \right)^3 \left(\frac{3}{5} \frac{P_f^2}{2m} \right) \cdot V$$

$$\left(\frac{2\pi^2 N}{g V} \right)^{1/3} = \frac{P_f}{\hbar}$$

$$E = V C_E P_f^5$$

$$N = V C_N P_f^3$$

Then

$$\Delta E_{\text{Tot}} = \Delta E_P + \Delta E_N = 2 \left[\overbrace{\frac{1}{2} \frac{\partial^2 E}{\partial p_f^2} (\Delta p_f)^2}^{\Delta E_P} + \overbrace{\frac{1}{2} \frac{\partial^2 E}{\partial p_f^2} (\Delta p_f)^2}^{\Delta E_N} \right]$$
$$= \frac{\partial^2 E}{\partial p_f^2} (\Delta p_f)^2$$

$$E = V \cdot C_E p_f^5$$

$$C_E = \frac{g}{2\pi^2} \frac{1}{5} \cdot \frac{1}{2m} \frac{1}{h^3}$$

$$\frac{\partial^2 E}{\partial p_f^2} = 5 \cdot 4 \cdot V C_E p_f^3$$

$$\frac{\Delta N}{2} = \frac{\partial N}{\partial p_f} \Delta p_f = 3C_N \cdot V p_f^2 \Delta p_f$$

$$\Delta E_{\text{Tot}} = 20 C_E \cdot V p_f^3 \cdot \left[\frac{\Delta N}{(3C_N \cdot V \cdot p_f^2)} \right]^2$$

$$\Delta E_{\text{Tot}} = \frac{20 C_E \cdot V}{9 C_N^2 V^2} \frac{1}{p_f} (\Delta N)^2 \quad p_f =$$

$$\Delta E_{\text{Tot}} = \frac{20 C_E}{9 C_N^2} \frac{1}{(p_f)} \frac{(\Delta N)^2}{V}$$

↘ Const

$$\Delta N = \left(\frac{N-Z}{2} \right)$$

$$V = \frac{4\pi}{3} R^3$$

$$V = \rho_0 \frac{A}{\rho_0}$$

$$\text{So - } \Delta E_{\text{TOT}} = \frac{20}{9} \frac{C_E}{C_N^2} \frac{1}{\rho_f} \frac{1}{\rho_0} \frac{1}{4} \cdot \frac{(N-Z)^2}{\frac{A}{\rho_0}}$$

$$C_N \rho_f^3 = \frac{\rho_0}{2} \rightarrow$$

$$\Delta E_{\text{TOT}} = \frac{20}{9} \frac{C_E}{C_N^2} \cdot \frac{1}{\rho_f} \cdot \frac{1}{4} \overbrace{\rho_0}^{2C_N \rho_f^3} \frac{(N-Z)^2}{A}$$

$$\Delta E_{\text{TOT}} = \frac{10}{9} \frac{C_E}{C_N} \rho_f^2 \left(\frac{N-Z}{A} \right)^2$$

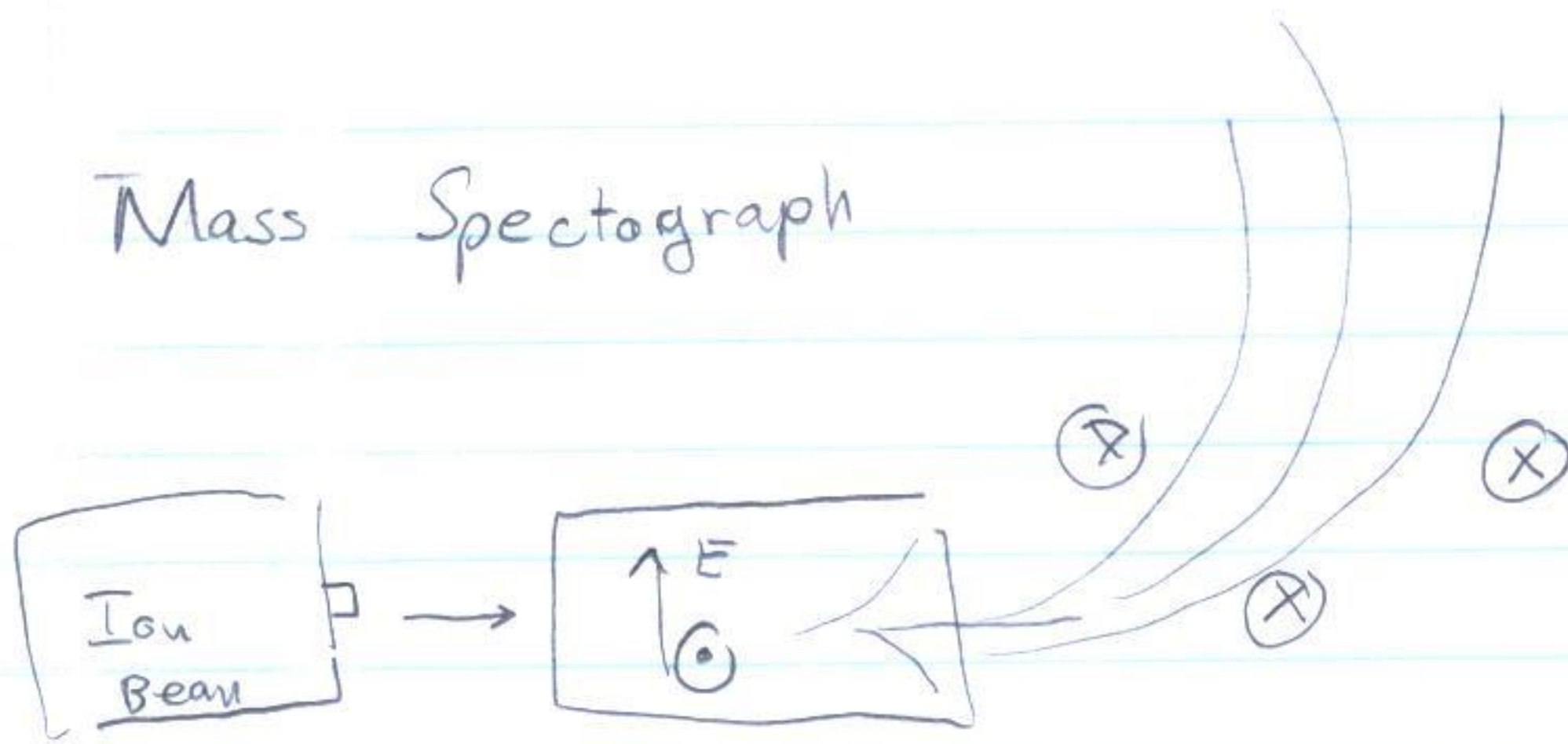
$$\Delta E_{\text{TOT}} = \frac{10}{9} \underbrace{\frac{\rho_f^2}{2m}} \left(\frac{N-Z}{A} \right)^2$$

~ 25 MeV

from before

$$\Delta E \sim 27 \text{ MeV} \left(\frac{N-Z}{A} \right)^2$$

Mass Spectrograph



$$F = qE \quad \text{up}$$

$$F = qvB \quad \text{down}$$

Straight Line :

$$qE = qvB$$

$$E = vB$$

Now we enter a magnetic Field :

$$m \frac{v^2}{R} = qBv$$

$$R = \frac{mv}{qB}$$

$$m = \frac{qRB^2}{v}$$