

Eq. (2.20) establishes the temperature dependence of the total radiation intensity: (see eq (2.10))

$$(2.21) \quad S(T) = \frac{c}{4} u(T) \equiv \sigma_{SB} T^4$$

Here, we have defined the Stefan-Boltzmann ~~law~~ constant

$$(2.22) \quad \begin{aligned} \sigma_{SB} &= 5.67 \cdot 10^{-5} \frac{\text{erg}}{\text{cm}^2 \text{s K}^4} \\ &= 5.67 \cdot 10^{-8} \frac{\text{W}}{\text{m}^2 \text{K}^4} \end{aligned}$$

II.4 Wien's displacement law

We have seen from the previous section that applying classical thermodynamic reasoning to the blackbody radiation, one can derive the functional dependence of $u(T)$.

Wien considered blackbody radiation under adiabatic compression. He found

$$(2.23) \quad \begin{aligned} u(f, T) df &= \cancel{f^3} g\left(\frac{f}{T}\right) \cancel{df} && \text{Wien's law (1893)} \\ &= f g\left(\frac{f}{T}\right) df && g = \text{arbitrary function} \end{aligned}$$

Since $f = \frac{c}{\lambda}$, it follows immediately that the spectral energy density distribution has a maximum at a wavelength λ_{\max} inversely proportional to T

$$(2.24) \quad \lambda_{\max} \cdot T = \text{const} = 0.29 \text{ cm} \cdot \text{K}$$

Wien's displacement law

II.5 Rayleigh-Jeans Formula; Ultraviolet Catastrophe

From classical electrodynamics, we know

Kirchhoff: $u(f, T)$ is universal function

Stefan-Boltzmann: $S(T) = \sigma_{SB} T^4$

Wien: a.) $u(f, T) = f^3 g\left(\frac{f}{T}\right)$

b.) $\lambda_{max} \cdot T = \text{const.}$

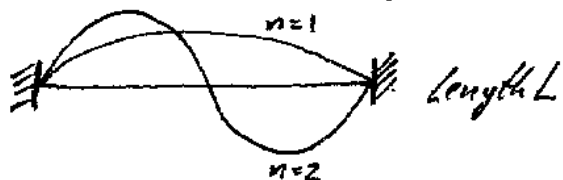
Now, we want to know the spectral distribution of u (i.e. f -dependence)

To this end: A: calculate the number of modes between f and $f+df$

B: to obtain $u(f, T)$, multiply by average energy per mode and divide by volume

A. How many modes fit into a cavity?

Waves have vanishing amplitudes at wall if they are static



$$(2.25) \quad L = n \cdot \frac{\lambda}{2}, \quad n = 1, 2, 3, \dots \text{ integer}$$

$$(2.26) \quad \equiv n \cdot \frac{\pi}{k}$$

$$(2.27) \quad k \equiv \frac{2\pi}{\lambda} \text{ is called 'wave vector'}$$

The n -th wave vector $k_n = n \frac{\pi}{L}$ characterizes the n -th mode

$$(2.28) \quad \delta k = k_{n+1} - k_n = \frac{\pi}{L} \quad \text{separation between modes in } k\text{-space}$$

Typically, one allows for the range $-\infty < k < \infty$ rather than $0 < k < \infty$ (amounts to multiplication by factor $1/2$).

The number of modes in interval dk is then

$$(2.29) \quad dN = \frac{1}{2} \frac{dk}{\delta k} = \frac{L}{2\pi} dk$$

In 3 dimensions, the same argument leads to

$$(2.30) \quad dN(\vec{k}) = dN_x dN_y dN_z = \underbrace{L_x L_y L_z}_{\text{Volume}} \frac{1}{(2\pi)^3} dk_x dk_y dk_z$$

$$= V \frac{d^3k}{(2\pi)^3}$$

The 'volume' in k -space is

$$(2.31) \quad d^3k = 4\pi k^2 dk$$



In the vacuum

$$(2.32) \quad |\vec{k}| = \frac{2\pi}{\lambda} = \frac{2\pi f}{c}$$

Inserting into (2.30)

$$(2.33) \quad dN(\vec{k}) = V \frac{4\pi k^2 dk}{(2\pi)^3} = V \frac{4\pi}{(2\pi)^3} \left(\frac{2\pi f}{c}\right)^2 \frac{2\pi}{c} df = V \frac{4\pi}{c^3} f^2 df$$

Each imag wave comes with 2 polarizations. Thus

$$(2.34) \quad dn(f) = \frac{8\pi}{c^3} f^2 df \quad \begin{array}{l} \text{density of modes} \\ \text{between } f \text{ and } f+df \end{array}$$

B. What is the average energy per wave mode?

Rayleigh-Jeans start from classical thermodynamics:

If a mode has energy E above the lowest lying energy E_0 , then it is found in a system of ~~one~~ temperature T with the probability

$$(2.35) \quad P(E) = \mathcal{N} e^{-(E-E_0)/k_B T}, \quad \mathcal{N} = \text{normalization} = P(E_0)$$

where the Boltzmann constant is

$$(2.36) \quad k_B = 1.38 \cdot 10^{-23} \frac{\text{J}}{\text{K}} = 8.62 \cdot 10^{-5} \frac{\text{eV}}{\text{K}}$$

The average energy \bar{E}_{Rj} of a mode in the blackbody spectrum is then

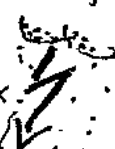
$$(2.37) \quad \bar{E}_{Rj} = \frac{\int_0^\infty dE E P(E)}{\int_0^\infty dE P(E)} = \frac{\int_0^\infty dE E e^{-\frac{E}{k_B T}}}{\int_0^\infty dE e^{-\frac{E}{k_B T}}} = \underline{\underline{k_B T}}$$

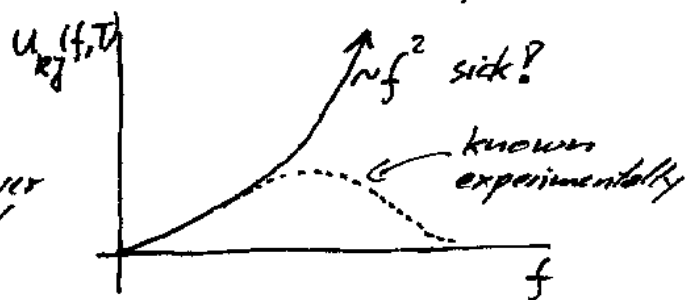
The spectral energy density $u(f, T)$ of blackbody radiation is then

$$(2.38) \quad \begin{aligned} u_{Rj}(f, T) &= \bar{E}_{Rj} dn(f) \\ &= \underline{\underline{\frac{8\pi}{c^3} f^2 k_B T df}} \end{aligned}$$

This has a sick ultra violet behavior (UV catastrophe)

$$(2.39) \quad u_{Rj}(T) = \int_0^\infty u_{Rj}(f, T) df$$

= infinity
 cloud over classical physics



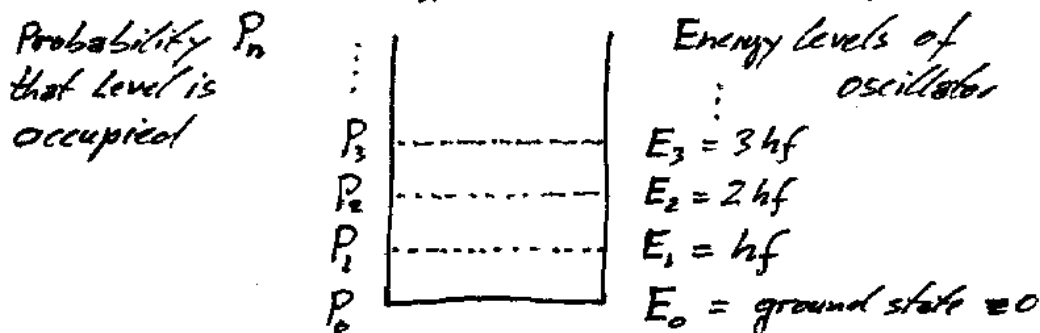
II.6 Planck's solution of the UV catastrophe

Properties of wall of the blackbody cavity can be chosen arbitrarily, since $u(f, T)$ is universal.

Planck views the wall as a collection of oscillators, which are in thermodynamic equilibrium with electro. field in cavity.

Planck assumes that for each frequency f in the cavity, the oscillators can emit or absorb only integer multiples of an elementary energy unit hf .

That means, the energy of the oscillator is quantized



We write again the Boltzmann distribution, which is based on classical statistical mechanics (2.35)

$$(2.40) \quad P_n = P_0 \exp\left[-\frac{E_n}{k_B T}\right] ; \quad E_n = nhf \\ = P_0 x^n$$

where we introduced the shorthand

$$(2.41) \quad x = e^{-\frac{hf}{k_B T}}$$

The average energy in the mode of frequency f is now [compare with (2.37)]

$$(2.42) \quad \bar{E}_{\text{Planck}} = \frac{\sum_n E_n P_n}{\sum_n P_n} = hf \frac{\sum_n n x^n}{\sum_n x^n}$$

To calculate (2.42), determine first the norm

$$S(x) \equiv \sum_{n=0}^{\infty} x^n = 1 + x + x^2 + \dots$$

$$\Rightarrow -x S(x) = -x - x^2 - x^3 - \dots$$

$$\Rightarrow S(x) - x S(x) = 1$$

$$= S(x) (1-x)$$

$$(2.43) \quad \Rightarrow \quad S(x) = \sum_{n=0}^{\infty} x^n = \frac{1}{1-x}$$

Now comes the numerator of (2.42)

$$(2.44) \quad \sum_{n=0}^{\infty} n x^n = x \sum_{n=0}^{\infty} n x^{n-1} = x \frac{d}{dx} \sum_{n=0}^{\infty} x^n = x \frac{d}{dx} S(x)$$

$$= x \frac{d}{dx} \frac{1}{1-x}$$

$$= x \frac{+1}{(1-x)^2}$$

Inserting (2.43) and (2.44) into (2.42), we find

$$(2.45) \quad \bar{E}_{\text{Planck}} = hf \frac{\sum_n n x^n}{\sum_n x^n} = hf \frac{\frac{x}{(1-x)^2}}{\frac{1}{1-x}} = hf \frac{x}{1-x}$$

$$= hf \frac{1}{\frac{1}{x} - 1}$$

$$= \frac{hf}{e^{\frac{hf}{k_B T}} - 1}$$

$h = \text{Planck's constant}$

$$(2.46) \quad \hbar \equiv \frac{h}{2\pi} = 1.054 \cdot 10^{-27} \text{ erg}\cdot\text{s} = 1.054 \cdot 10^{-34} \text{ J}\cdot\text{s}$$