

You can gain a total of 24 pts. Good luck!

1. (8 pts) Apply your knowledge of one-dimensional quantum mechanics to the flux of particles of mass m and energy $E > 0$, incident from the left on a potential step of the form

$$\begin{aligned} V(x) &= 0, & \text{for } x < 0, \\ V(x) &= -V, & \text{for } x > 0. \end{aligned}$$

- (a) Write down the Schrödinger equation for this problem.
- (b) Find the solutions of the Schrödinger equation for $x > 0$ and $x < 0$. Then write down the continuity conditions.
- (c) Calculate the probability with which the particles will be reflected (reflection coefficient). What would happen classically?
2. (4 pts) A particle of mass m moving along the x -axis with a velocity component $+u$ collides head-on and sticks to a particle of mass $m/3$ moving along the x axis with the velocity component $-u$. Write down energy and momentum conservation for this system. Calculate the mass M of the resulting particle?
3. (4 pts) The pion has an average lifetime of $t_0 = 26$ ns when at rest. For it to travel $\Delta x = 10.0$ m on average, how fast must it move? Give first the formula in terms of velocity v , distance Δx and lifetime t_0 , then put in numbers.
4. (4 pts) The z -component of the angular momentum operator takes the form

$$\hat{L}_z = -i\hbar \left(x \frac{\partial}{\partial y} - y \frac{\partial}{\partial x} \right). \quad (1)$$

- (a) Calculate the commutator $[\hat{L}_z, \hat{y}]$ between the position operator \hat{y} in the y -direction and \hat{L}_z .
- (b) Answer without calculation the question: For the wave functions Ψ_{n,l,m_l} characterizing the bound states of the electron in the hydrogen atom, what is $\hat{L}_z \Psi_{n,l,m_l}$?
5. (4 pts) PHY251-Trivial pursuit: answer in one line without calculation
- (a) How do the energy levels E_n depend on the principal quantum number n for a) hydrogen atom, b) infinitely deep square well, c) harmonic oscillator?
- (b) For a hydrogen atom with orbital quantum number $l = 2$, what are the allowed magnetic quantum numbers?
- (c) space-quantization: Consider the total angular momentum L of the hydrogen atom, and its z -component L_z . What are the possible orientations of the total angular momentum with respect to an arbitrary z -direction?
- (d) State Bohr's correspondence principle in one sentence (not more!).